

Synthesis of Fluorinated Carbocyclic Nucleosides: Preparation of (\pm)-Carbocyclic-FMAU and Some Congeners

Keith Biggadike,^a Alan D. Borthwick,^a Derek Evans,^a Anne M. Exall,^a Barrie E. Kirk,^a Stanley M. Roberts,^{a†} Les Stephenson,^a Peter Youds,^a Alexandra M. Z. Slawin,^b and David J. Williams^b

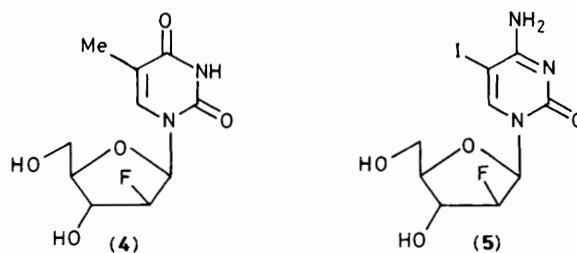
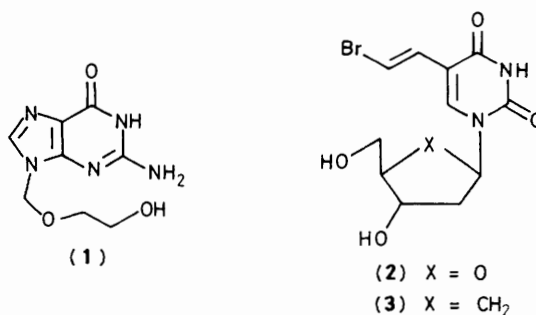
^a Department of Microbiological Chemistry, Glaxo Group Research, Greenford, Middlesex UB6 0HE, U.K.

^b Department of Chemistry, Imperial College of Science and Technology, South Kensington, London SW7 2AY, U.K.

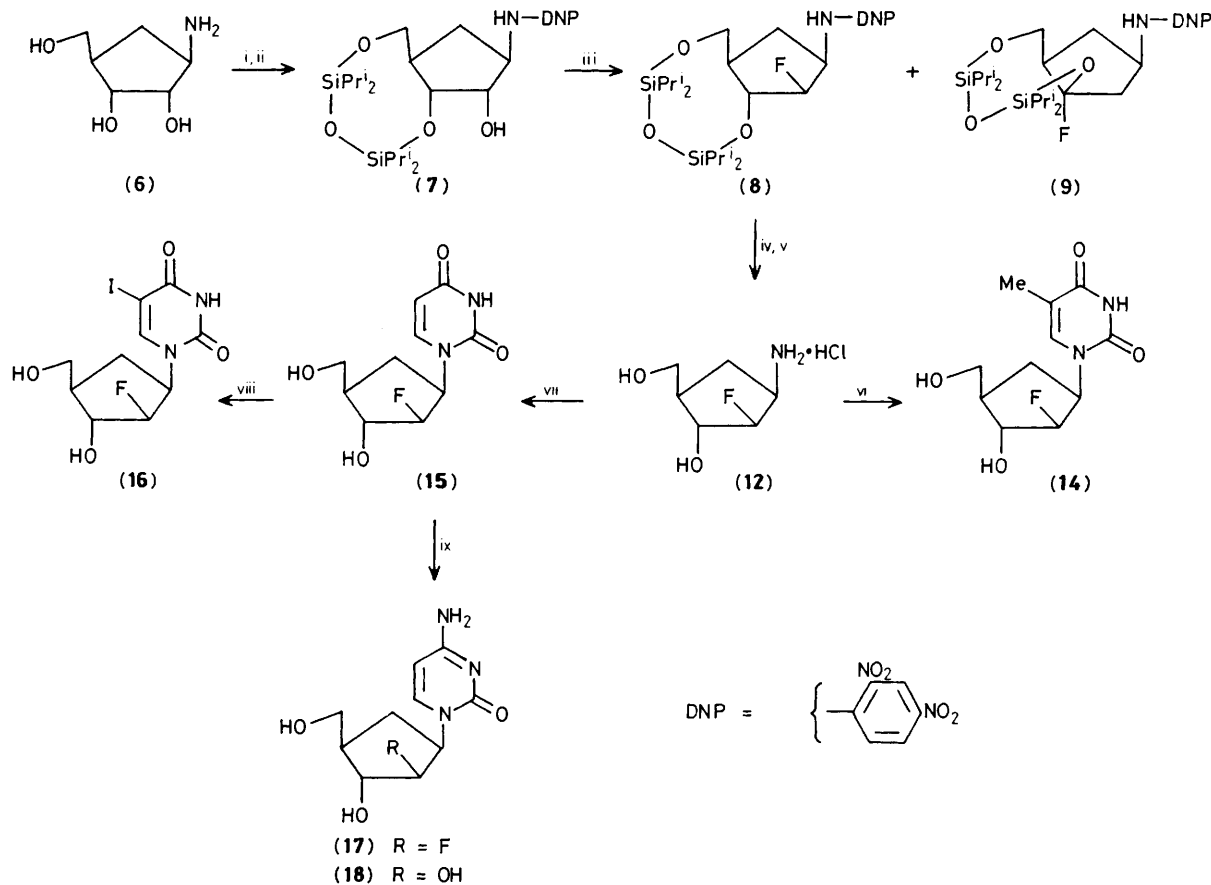
The alcohols (7) and (19) and the ketone (28) were treated with diethylaminosulphur trifluoride (DAST) to give the fluoro-compounds (8), (21), and (29) respectively: compound (8) was converted into the potential anti-viral agents carbocyclic-FMAU (14) and carbocyclic-FIAU (16) while compound (21) afforded the 2'- α -fluoro-carbocyclic nucleosides (24), (25), (27), and compound (29) gave the difluoro-analogue (32) [crystal data were obtained on compounds (8) and (9)].

One of the major areas of interest in modern-day medicinal chemistry involves the search for anti-viral agents.¹ In particular, the need for novel, orally active agents for the treatment of herpes simplex virus (HSV) infections is of paramount importance.² While acyclovir (1) is, at present, the compound of choice for use in the clinic against infections caused by HSV-1 and HSV-2,³ more potent anti-herpes compounds such as 5-(2-bromovinyl)-2'-deoxyuridine (BVDU) (2), 1-(2'-deoxy-2'-fluoro-1'- β -D-arabinofuranosyl)-5-methyluracil (FMAU) (4), and 1-(2'-deoxy-2'-fluoro-1'- β -D-arabinofuranosyl)-5-iodocytosine (FIAC) (5) have attracted considerable attention.⁴

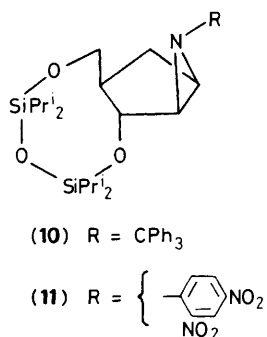
In an effort to improve the pharmacokinetics of the sugar derivative BVDU, the carbocyclic analogue (3) was prepared.⁵ However both BVDU (2) and the carbocyclic analogue (3) show only weak activity against HSV-2, so despite the improved bioavailability of the latter compound, it is only poorly effective against HSV-2 infections *in vivo*. With this background knowledge, we took up the challenge to make



† Present address: Department of Chemistry, University of Exeter, Exeter EX4 4QD, U.K.



Scheme 1. Reagents: i, DNP-F, dimethylformamide (DMF), Na_2CO_3 , room temp.; ii, $\text{O}(\text{Pr}_2\text{SiCl})_2$, DMF, imidazole; iii, DAST, CH_2Cl_2 , -30°C ; iv, $\text{Bu}_4\text{N}^+\text{F}^-$, tetrahydrofuran (THF); v, Amberlite IR 400 (OH^-), H_2O , Me_2CO ; vi, $\text{EtOCH}=\text{C}(\text{Me})\text{CONCO}$ (13), 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU), DMF, -20°C then 2 M-HCl, heat; vii, $\text{EtOCH}=\text{CHCONCO}$, DBU, DMF, -20°C then 2 M-HCl, heat; viii, I_2 , HNO_3 , CHCl_3 ; ix, Ac_2O , $\text{C}_5\text{H}_5\text{N}$ then *m*- $\text{ClC}_6\text{H}_4\text{-O-P}(\text{O})\text{Cl}_2$, $\text{C}_5\text{H}_5\text{N}$, triazole then NH_3 , MeOH, heat.



the (\pm)-carbocyclic counterpart of the broader spectrum anti-viral agent FMAU (4) (and some related compounds) and we report the success of this venture in this communication.

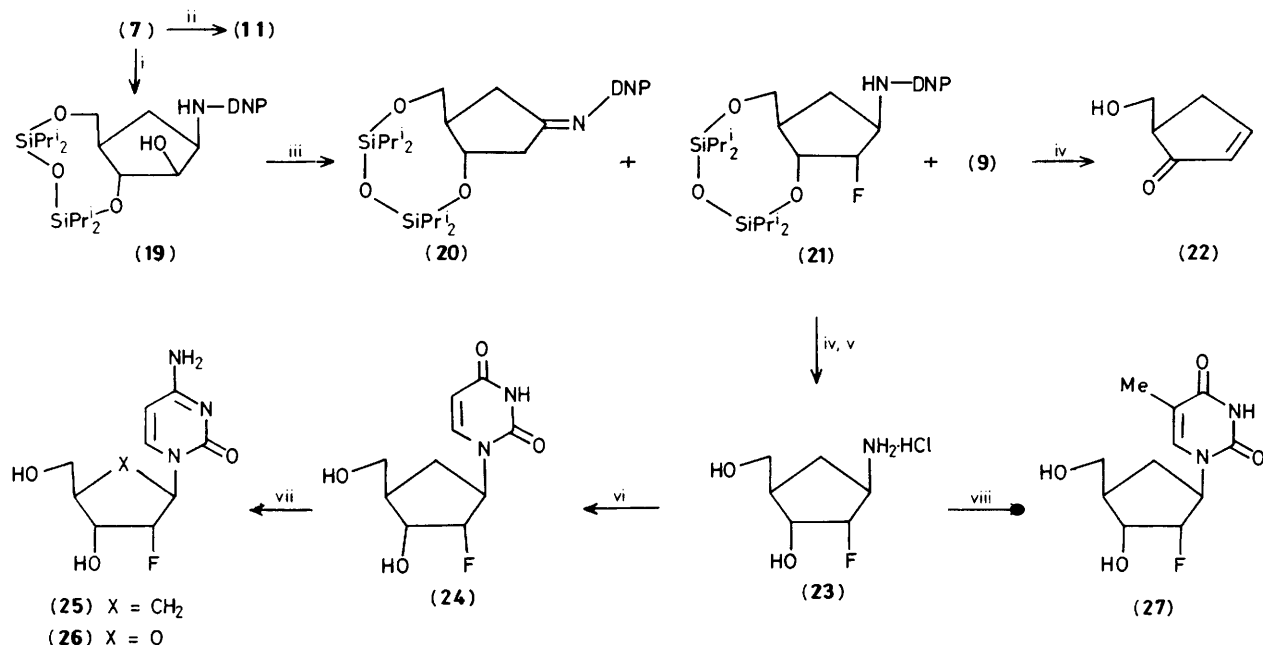
The (\pm)-amino-triol (6) is available from cyclopentadiene in seven steps.⁶ Protection of the amino-group with the 2,4-dinitrophenyl (DNP) moiety (89%) and formation of the oxybis(di-isopropylsilyl) (TIPS) derivative (79%) gave the key intermediate (7) (Scheme 1). Treatment of the alcohol (7) with diethylaminosulphur trifluoride (DAST)⁷ (2 equiv.) in dichloromethane at -30°C gave the desired fluoro-compound (8) (ca. 74%) and a small amount of the protected fluoro-hydrin (9) (ca. 18%) (*vide infra*). The configuration of the fluorine atom in (8) was confirmed by X-ray crystallography

(Figure 1).[‡] Thus the DAST reaction converts (7) into (8) with inversion of configuration at C-2';⁸ this conversion is complementary to other work which demonstrated the usefulness of this reagent in the synthesis of fluoro-sugars.^{9,10} Two points are worthy of note: first, if an amino-protecting group was employed which did not significantly reduce the electron

[‡] *Crystal data:* (\pm)-(8 α ,9 α)-8-[(2,4-Dinitrophenyl)amino]-9-fluorohexahydro-2,2,4,4-tetrakis(1-methylethyl)cyclopenta[*f*]-1,3,5,2,4-trioxadisilocene (8), $\text{C}_{24}\text{H}_{40}\text{FO}_7\text{N}_3\text{Si}_2$, $M = 557.8$, monoclinic, $a = 14.009(3)$, $b = 10.458(3)$, $c = 20.646(4)$ Å, $\beta = 101.03(2)^\circ$, $U = 2969$ Å³, space group $P2_1/a$, $Z = 4$, $D_c = 1.25$ g cm⁻³, $\mu(\text{Cu-K}\alpha) = 15$ cm⁻¹. The structure was solved by direct methods and refined anisotropically to give $R = 0.085$, $R_w = 0.063$ for 1707 independent observed reflections [$|F_o| > 3\sigma(|F_o|)$, $\theta \leq 50^\circ$].

(\pm)-(8 α ,10 β)-8-[(2,4-Dinitrophenyl)amino]-10-fluorohexahydro-2,2,4,4-tetrakis(1-methylethyl)-cyclopenta[*f*]-1,3,5,2,4-trioxadisilocene (9), $\text{C}_{24}\text{H}_{40}\text{FO}_7\text{N}_3\text{Si}_2$, $M = 557.8$, monoclinic, $a = 8.870(2)$, $b = 15.129(3)$, $c = 52.150(9)$ Å, $\beta = 122.03(2)^\circ$, $U = 5933$ Å³, space group $C2/c$, $Z = 8$, $D_c = 1.25$ g cm⁻³, $\mu(\text{Cu-K}\alpha) = 15$ cm⁻¹. The structure was solved by direct methods and refined anisotropically to give $R = 0.046$, $R_w = 0.047$ for 2512 independent observed reflections [$|F_o| > 3\sigma(|F_o|)$, $\theta \leq 50^\circ$].

In both cases data were measured on a Nicolet R3m diffractometer with $\text{Cu-K}\alpha$ radiation (graphite monochromator) using ω -scans. Atomic co-ordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.



Scheme 2. Reagents: i, (a) 4-*N,N*-dimethylaminopyridine (DMAP), CH₂Cl₂, MeSO₂Cl, (b) CsOCOMe, 18-crown-6, toluene, heat, (c) NaOMe, MeOH; ii, Ph₃P, EtO₂C-N=N-CO₂Et, *p*-MeC₆H₄SO₃Me, toluene; iii, DAST (1.5 equiv.), CH₂Cl₂, 0°C; iv, Buⁿ₄N⁺F⁻, THF; v, Amberlite IR 400 (OH⁻), H₂O, Me₂CO; vi, EtOCH=C(Me)CONCO, DBU, DMF, -20°C then 2 M-HCl, heat; vii, Ac₂O, C₅H₅N, DMAP then *m*-ClC₆H₄-O-P(O)Cl₂, C₅H₅N, triazole, then NH₄OH; viii, EtOCH=C(Me)CONCO (13), DBU, DMF, -20°C then 2 M-HCl, heat.

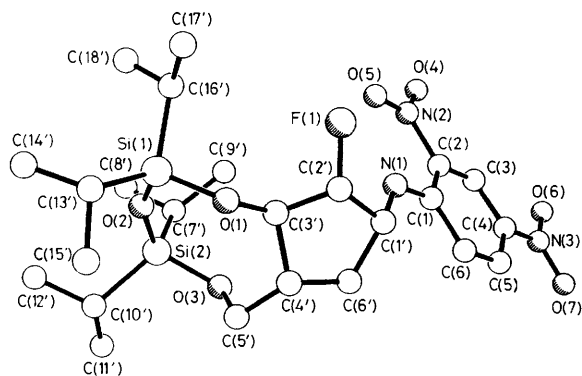


Figure 1. The molecular structure of (8). There is an intramolecular hydrogen bond between N(1) and O(5) [N(1) ··· O(5) 2.62, H(1) ··· O(5) 1.90 Å, ∠N(1)H(1)O(5) 130°].

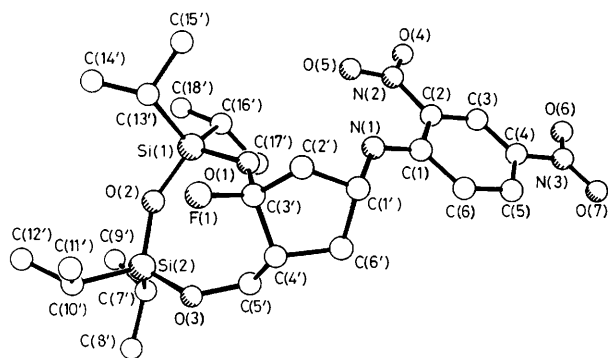


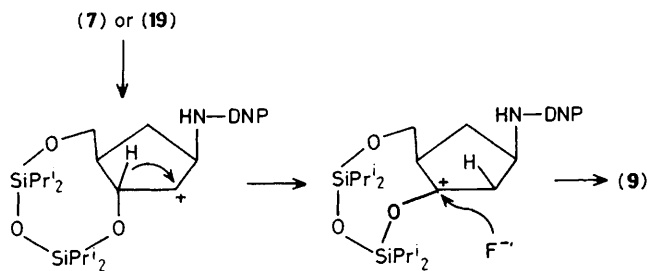
Figure 2. The molecular structure of (9). There is an intramolecular hydrogen bond between N(1) and O(5) [N(1) ··· O(5) 2.65, H(1) ··· O(5) 1.94 Å, ∠N(1)H(1)O(5) 129°].

density on the nitrogen atom (e.g. trityl), the only product isolable from the DAST reaction was the corresponding aziridine [e.g. (10)]. Secondly, the TIPS protecting group was stable under the reaction conditions, a fact that was not predictable from earlier reports.¹¹ The fluoro-compound (8) was deprotected to give the aminofluorodiol hydrochloride (12) (86%) and this compound was converted into carbocyclic-FMAU (14) (21%) using the isocyanate (13) followed by treatment with hydrochloric acid in the prescribed fashion (Scheme 1).¹¹

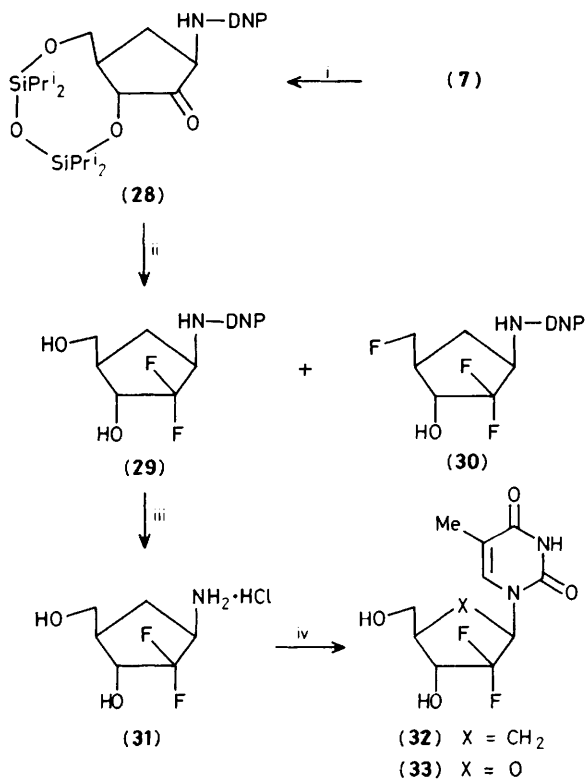
The salt (12) was converted into the uracil derivative (15) which gave access to the carbocyclic analogue (16) [20% from compound (12)] of 1-(2'-deoxy-2'-fluoro-1-β-D-arabinofuranosyl)-5-iodouracil (FlAU) and the carbocyclic analogue (17) [7.5% from compound (12)] of 1-(2'-deoxy-2'-fluoro-1-β-D-arabinofuranosyl)cytosine. Carbocyclic 1-β-arabinofuranosylcytosine (18) has been prepared previously.¹²

Inversion of the free hydroxy group in compound (7) was accomplished by methanesulfonylation (72%), displacement of the MeSO₃ group by acetate ion (95%), and de-esterification using sodium methoxide (98%) to give the required alcohol (19) (Scheme 2). [An attempted Mitsunobu reaction on alcohol (7) gave the aziridine (11).]

Treatment of the alcohol (19) with DAST gave the protected fluorohydrin (9), the imine (20), and the required fluoro-compound (21) (ca. 25% yield). The imine (20) is presumably formed by dehydration¹³ followed by rearrangement of the double bond under the reaction conditions. The structure of compound (9) was confirmed by X-ray crystallography (Figure 2):‡ one possible mechanism for the formation of (9) involves loss of the activated 2'-hydroxy group,¹⁴ and proton shift to give an oxygen-stabilized carbenium ion (Scheme 3) which is attacked by local fluoride ion from the less hindered face. As expected, treatment of (9) with tetrabutylammonium fluoride (TBAF) gave 5-hydroxymethylcyclopent-2-enone (22). The whole process (19) → (9) → (22) is



Scheme 3



Scheme 4. Reagents: i, Me_2SO , $(\text{CF}_3\text{CO})_2\text{O}$, CH_2Cl_2 -65°C to room temp. then Pr_2NEt ; ii, DAST, CH_2Cl_2 , room temp., then $\text{Bu}^n_4\text{N}^+\text{F}^-$, THF; iii, Amberlite IR 400 (OH^-), Me_2CO , H_2O then 2 M-HCl; iv, $\text{EtOCH}=\text{C}(\text{Me})\text{CONCO}$ (13), DBU, DMF, -20°C , then 2 M-HCl, heat.

reminiscent of a transformation described by Robins some years ago.¹⁵

The crude 2'- α -fluoro-compound (21) was deprotected in two steps to give the hydrochloride (23) [17% from (19)] and this compound was converted into the uracil derivative (24) [62% yield from (23)] and subsequently into the cytosine derivative (25) [49% yield from (24)], as well as carbocyclic 1-(2'-deoxy-2'- α -fluororibofuranosyl)-5-methyluracil (27) [34% yield from (23)] (Scheme 2).

The protected aminotriol (7) was oxidised under carefully controlled conditions to afford the ketone (28) (40%) (Scheme 4). Conversion of simple ketones into the corresponding difluoromethylene compounds using DAST is known to require harsh reaction conditions.¹⁶ Not surprisingly, therefore, the ketone (28) was converted into the

difluoro-compound (29) in variable, at best modest, yield (12–27%) using DAST under the prescribed conditions followed by TBAF to remove the silyl protecting group. A small amount of the trifluoro-compound (30) was isolated as a minor product from one of these reactions. Removal of the dinitrophenyl-protecting group from (29) gave the salt (31) (71%) which was converted into carbocyclic 1-(2'-deoxy-2'- α -difluororibofuranosyl)-5-methyluracil (32) (51%) using the isocyanate (13) followed by acid treatment.

Carbocyclic-FMAU (14) showed activity in the HSV-1 plaque reduction assay but at a level significantly lower (*ca.* 1000 fold) than that attained by FMAU (4). None of the compounds (24), (25), (27), and (32) was active against HSV-1 infected cells *in vitro* at a concentration of 300 $\mu\text{g}/\text{ml}$. In contrast the fluoro-sugars (26) and (33) have been reported to show activity against HSV-infected cells.¹⁷

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- Inversion of configuration is the usual course of events in such reactions (*e.g.* D. G. Drueckhammer and C.-H. Wong, *J. Org. Chem.*, 1985, **50**, 5912) but exceptional cases have been noted (see S. Castillon, A. Dessinges, R. Faghih, G. Lukacs, A. Olesker, and T. T. Thang, *ibid.*, 1985, **50**, 4913). The numbering of the carbocyclic nucleosides is in accord with the system used by G. C. Madhavan and J. C. Martin, *ibid.*, 1986, **51**, 1287.
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